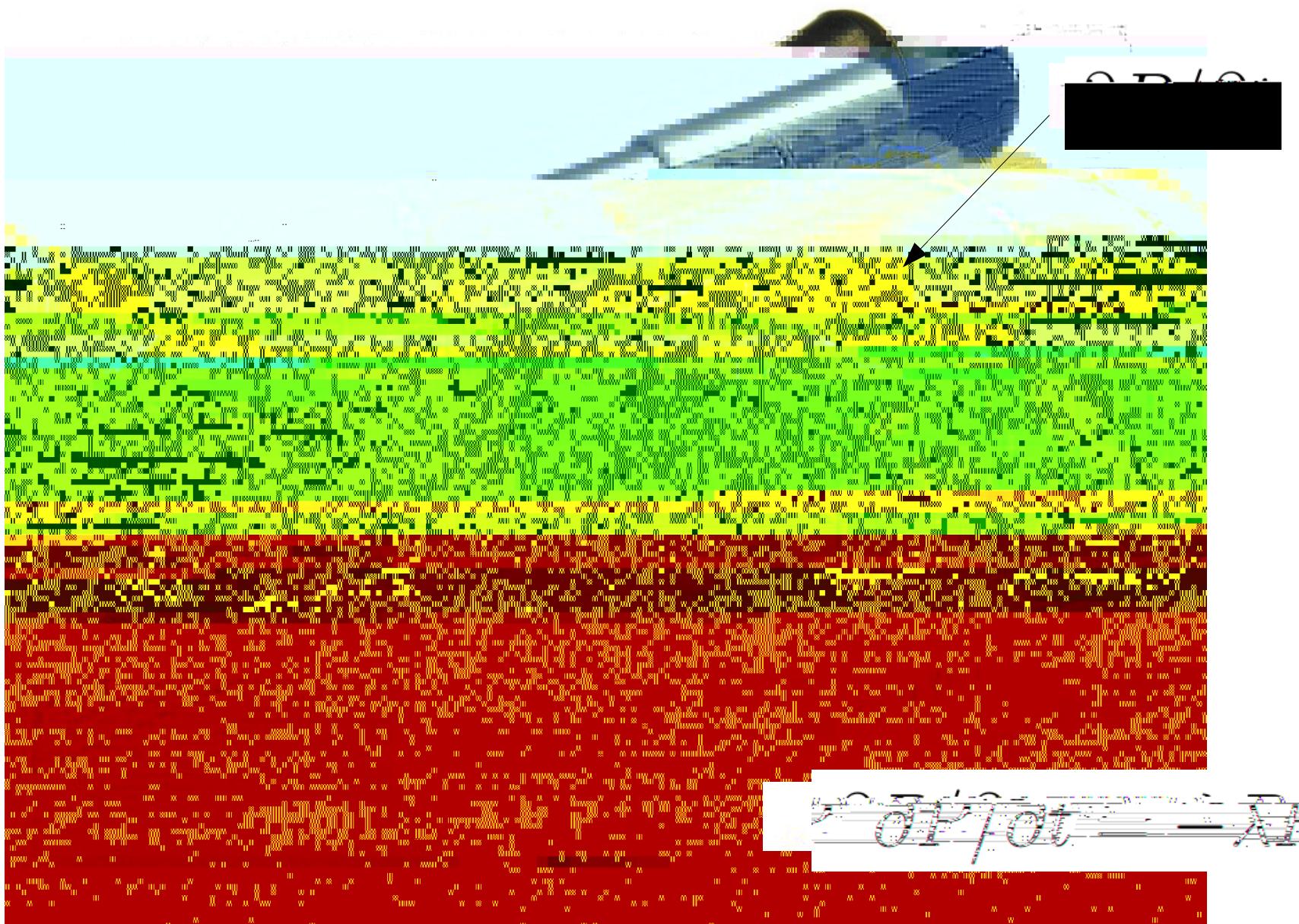


The rate of radioactive decay is ON ! a function of the amount present
and is not affected by pressure" temperature or chemistry



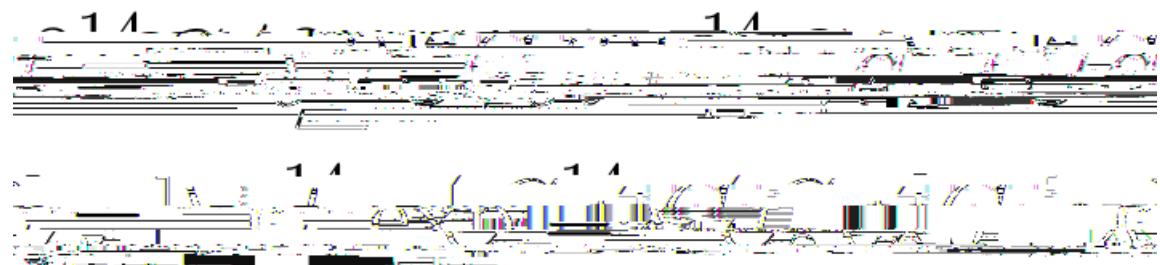
P

$$\frac{N(t) - N_0}{N_0} = -\lambda t$$

2.

3.

P



Inferring the initial amount of &

P

&



P

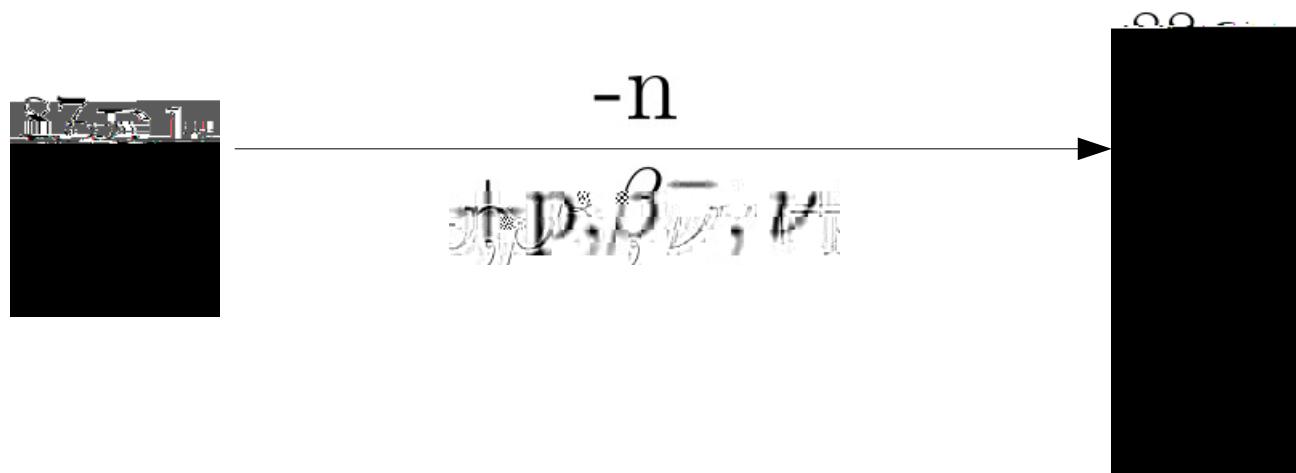
$$P = P_0 \exp[-\lambda t]$$

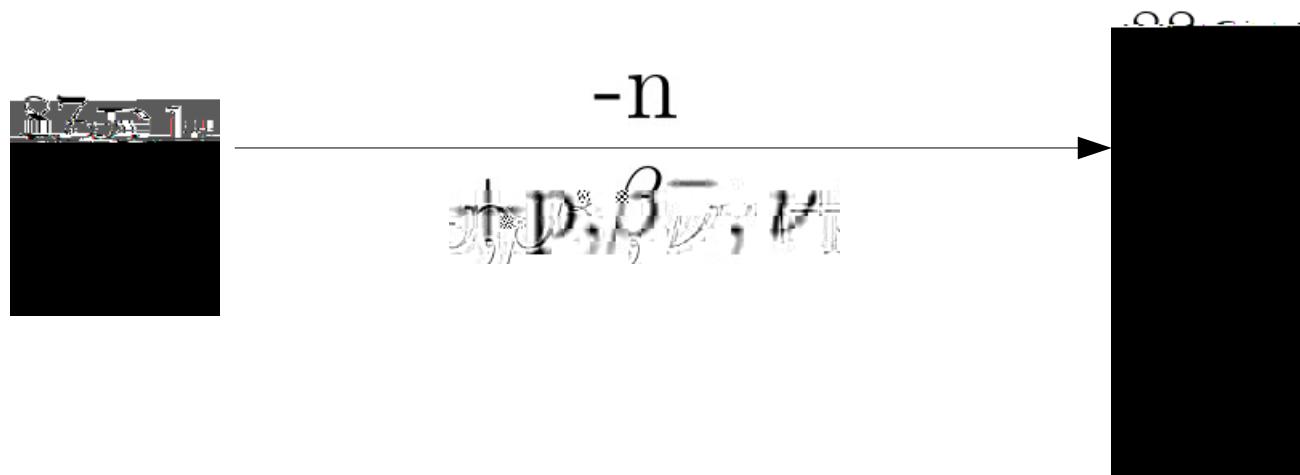
$$\lambda t] = P = P_0 (P + D) \exp[-\lambda t]$$

$$D + P = P \exp[\lambda t] + P = P$$

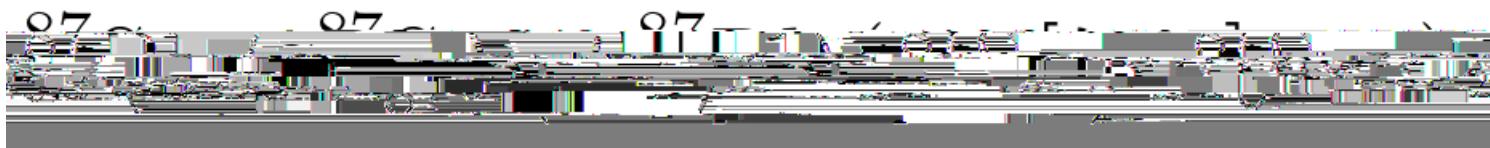
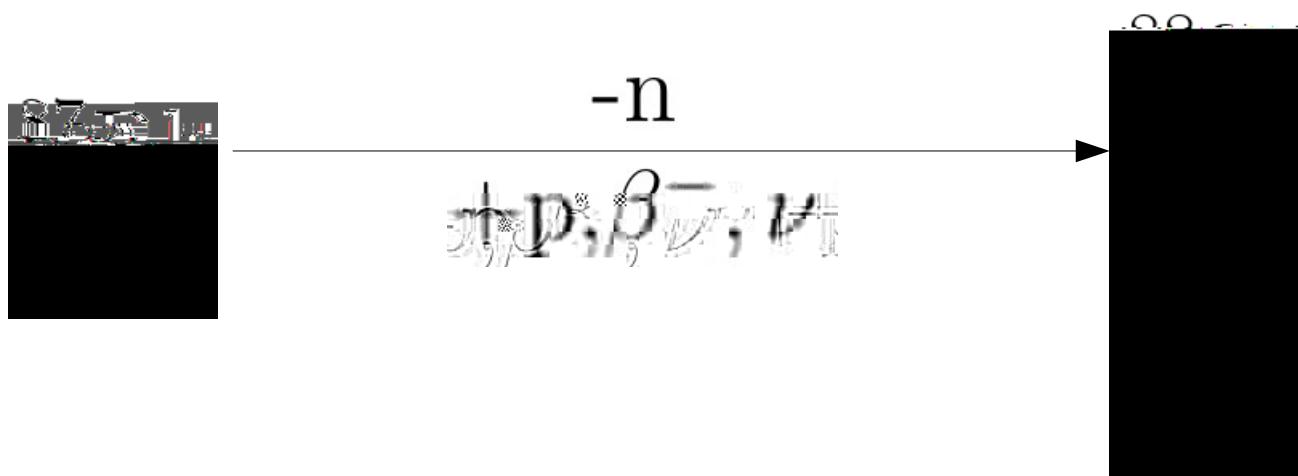
$$(exp[D] = P(\exp[\lambda t] - 1))$$

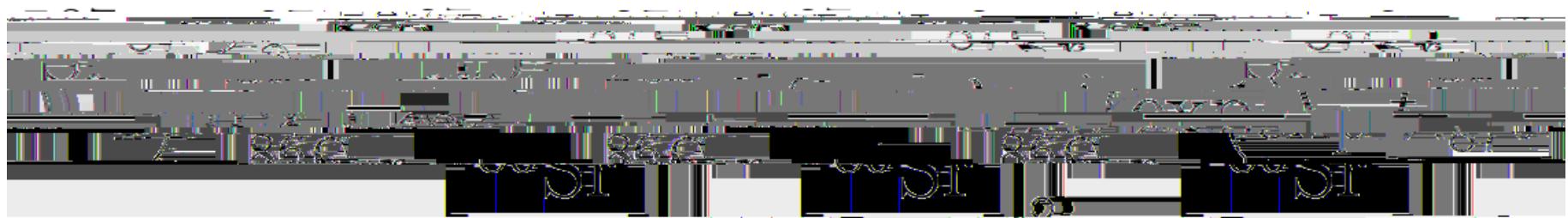
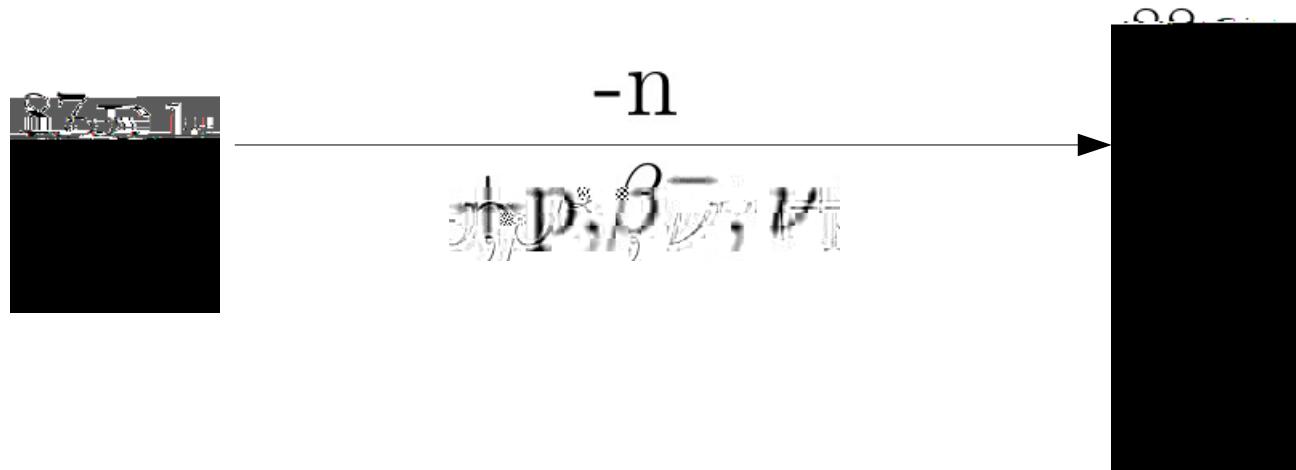
$$D = \ln[(1 + P)/P]/\lambda$$

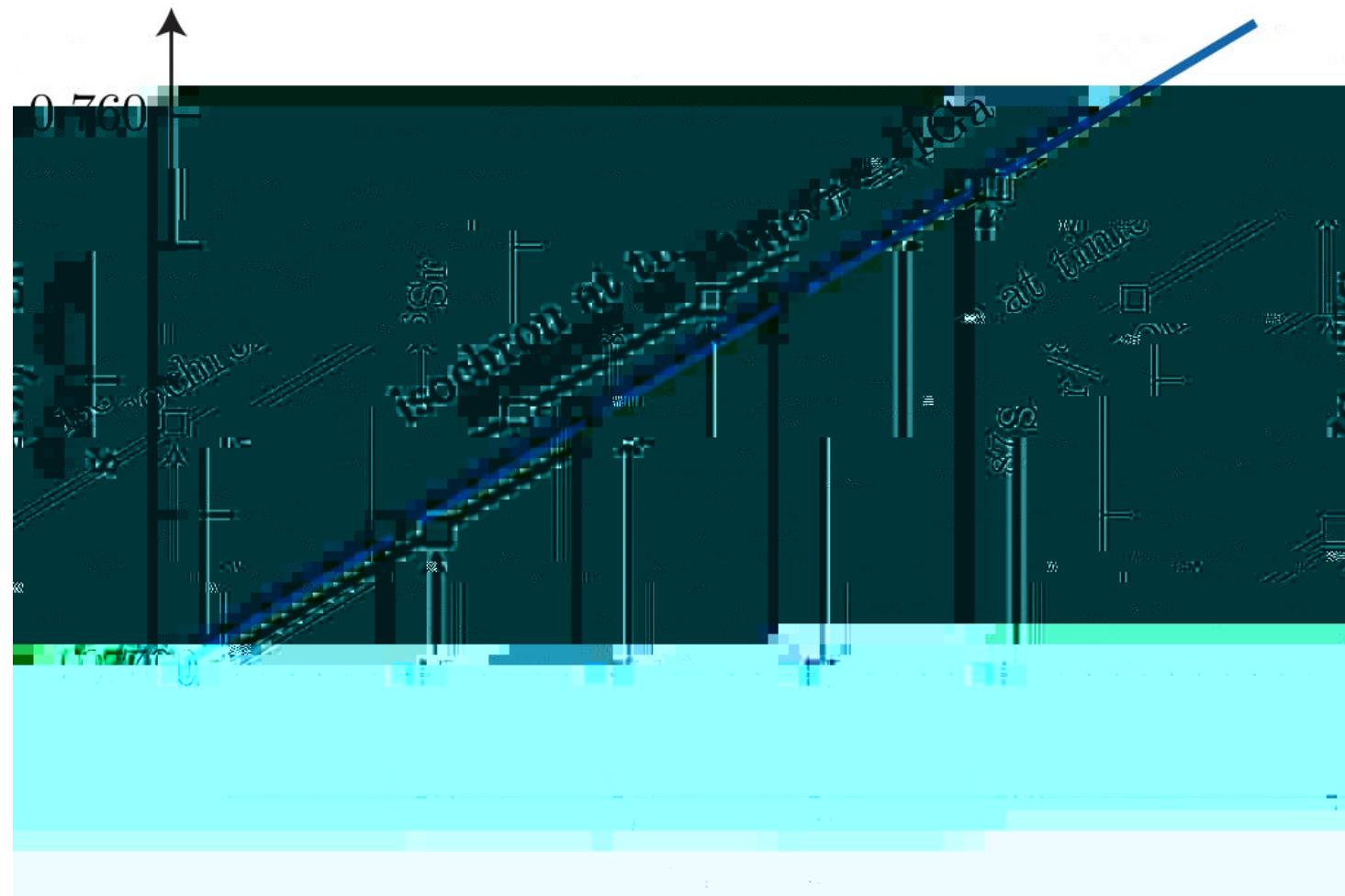




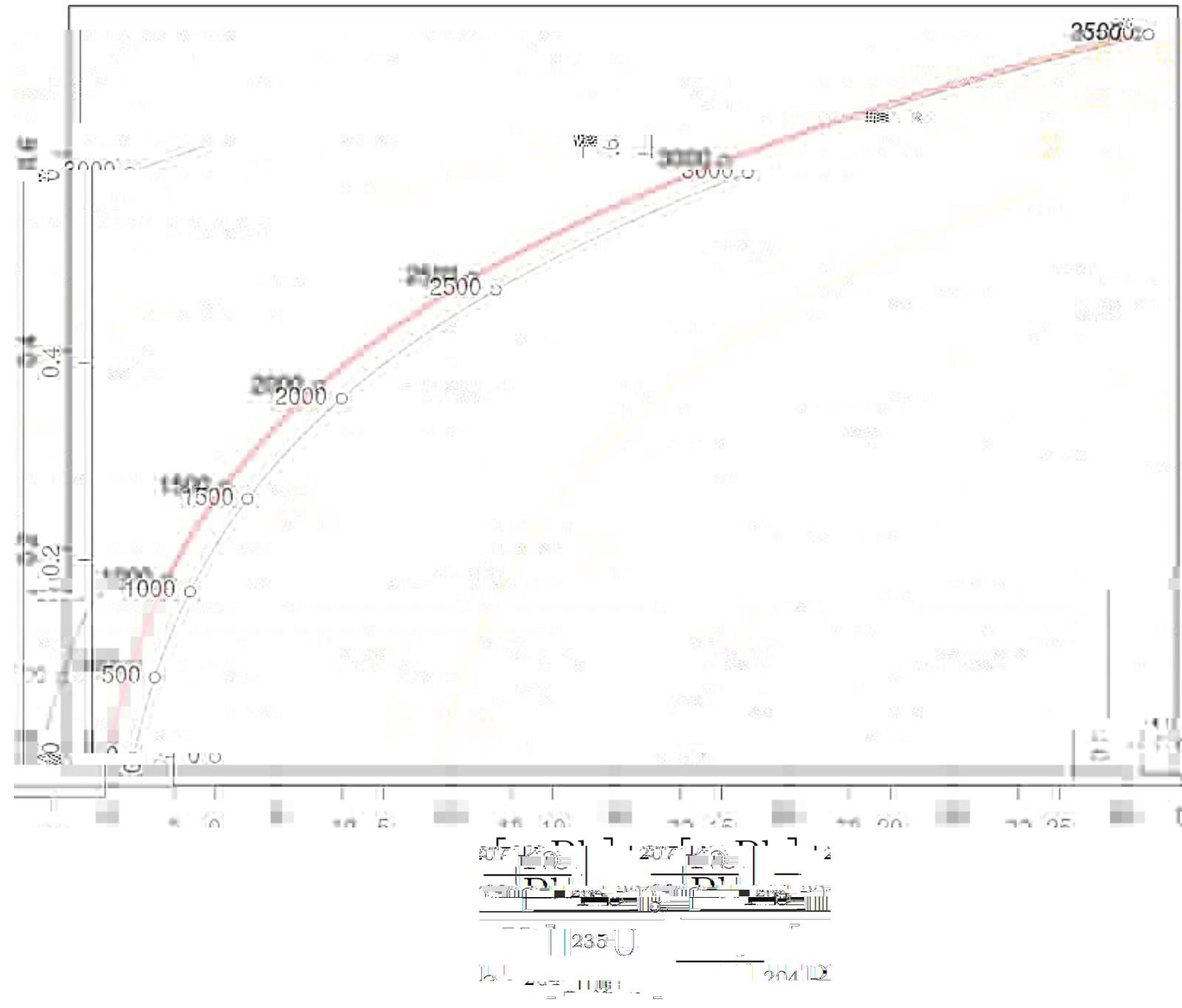
$${}^{87}\text{Sr} = {}^{87}\text{Rb}(\exp[\lambda_{87}t] - 1)$$

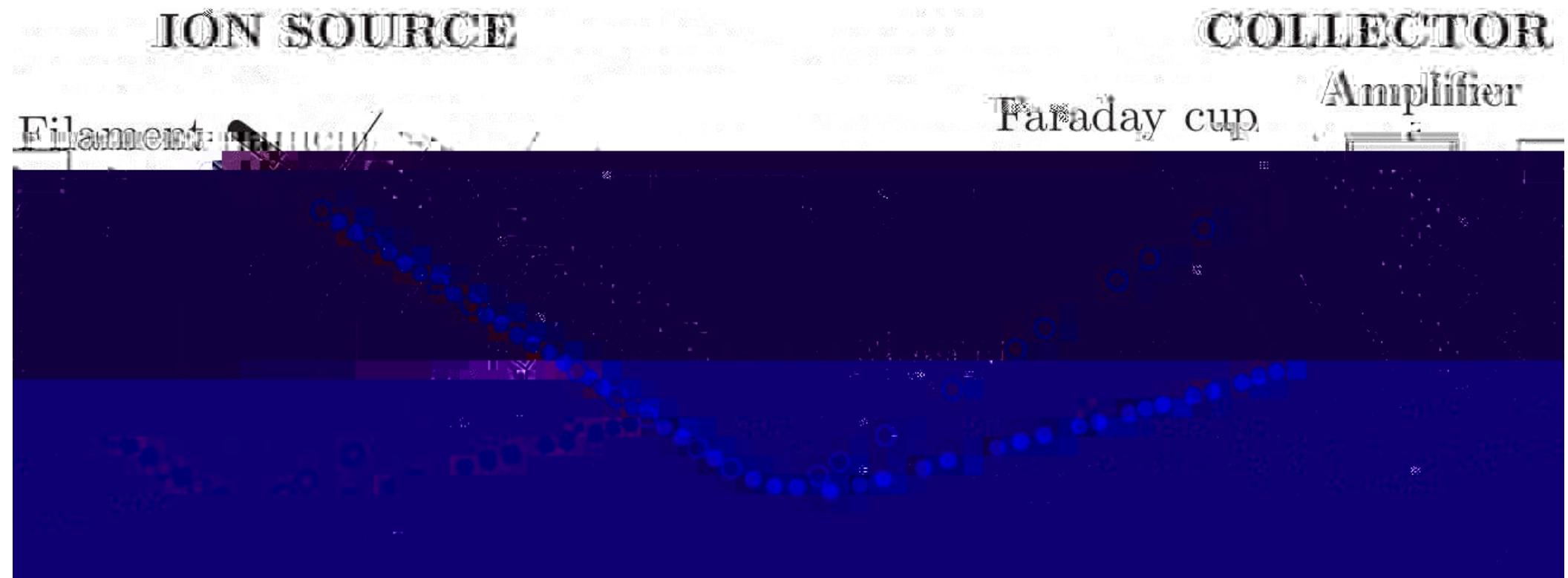


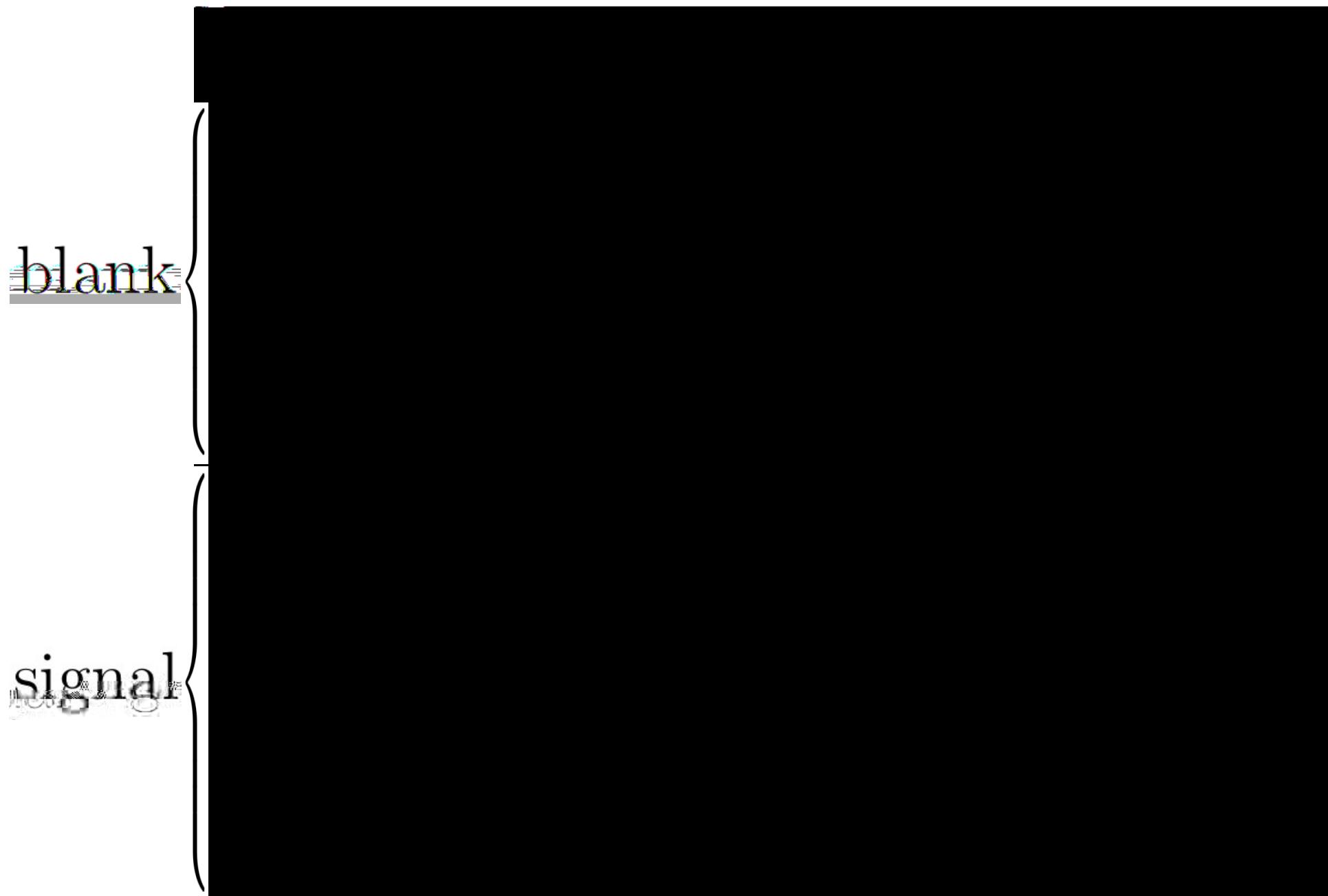










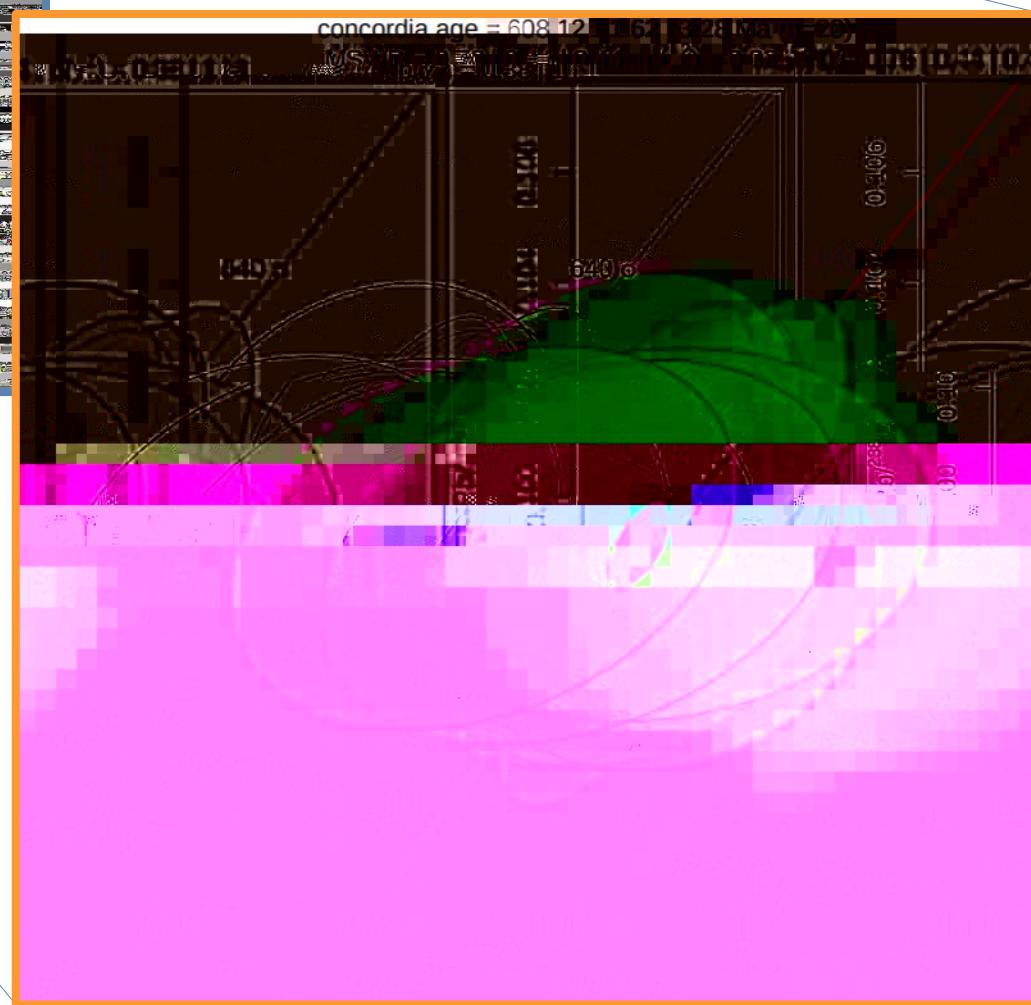


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signal

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signals





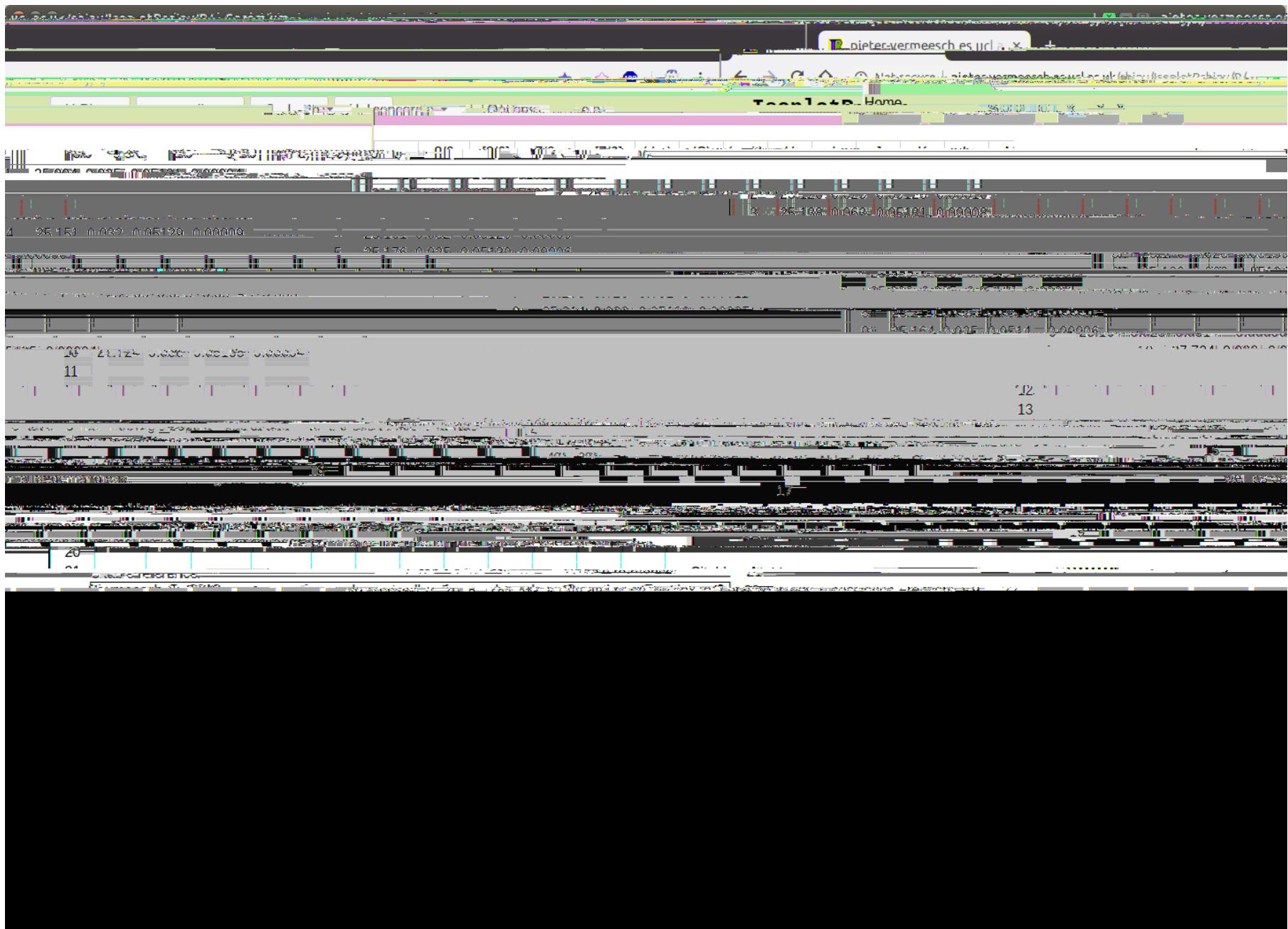
ET_Redux

IsoplotR

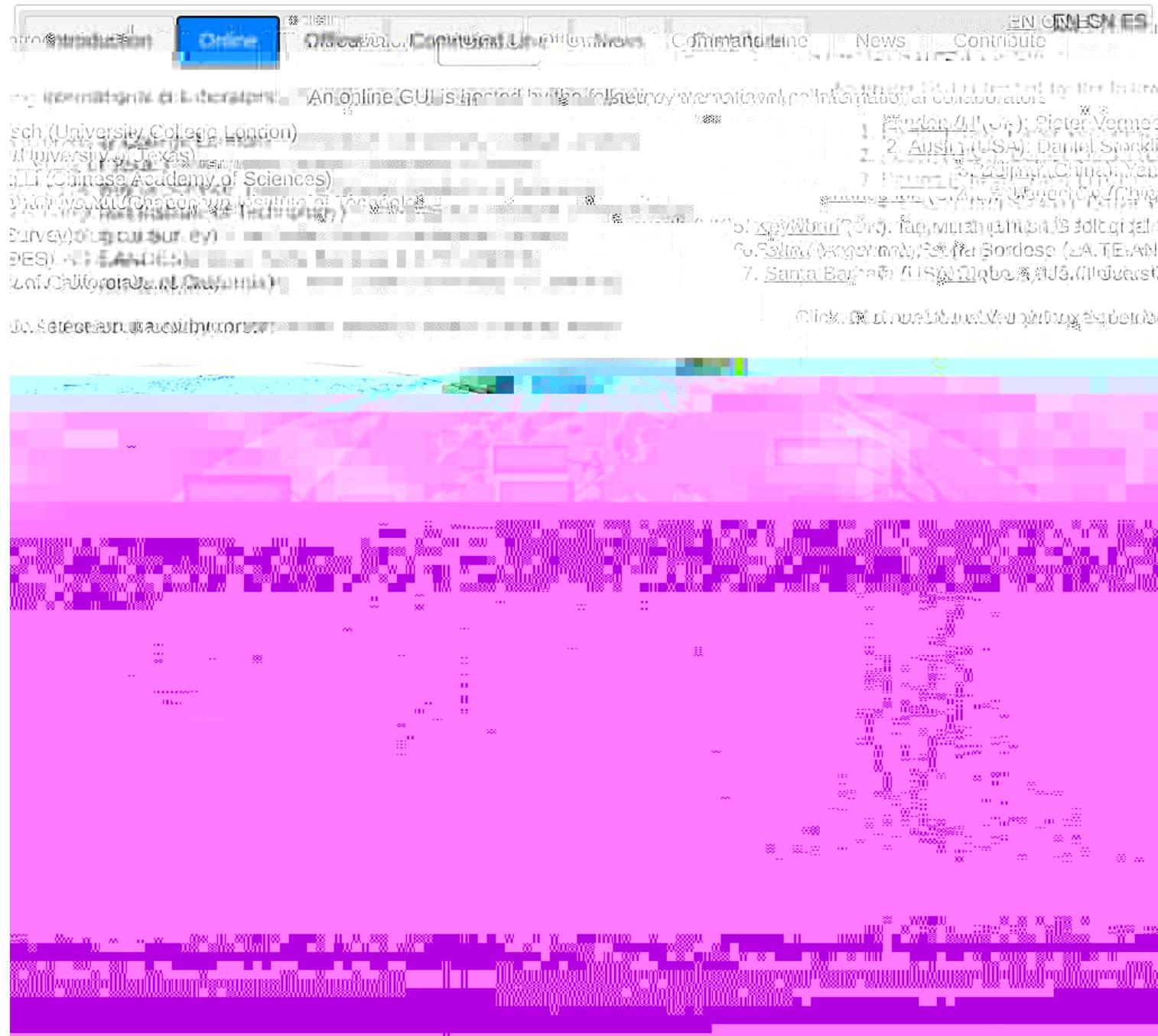
IsoplotRgui

PyChron

Ar-Ar_Redux





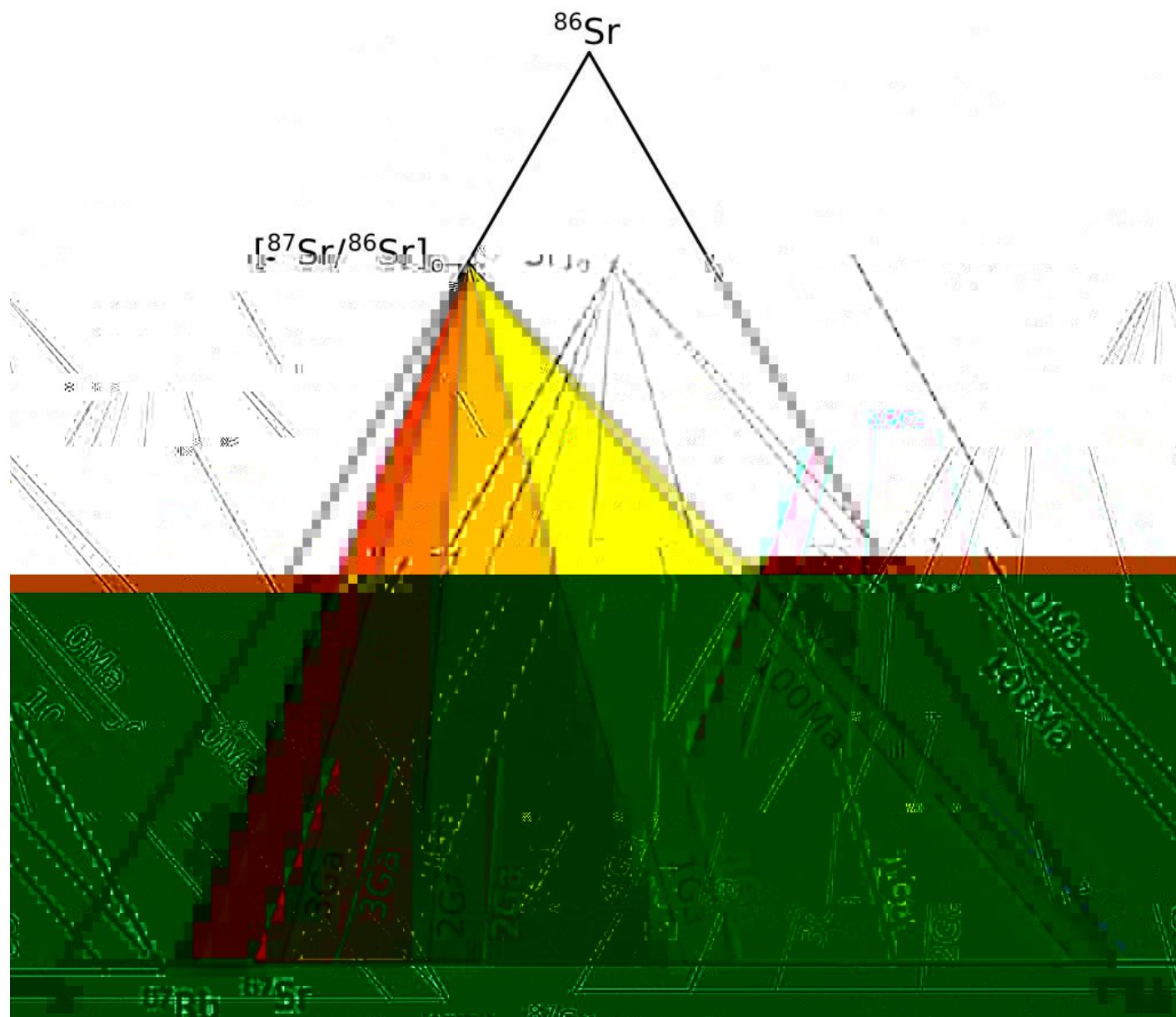


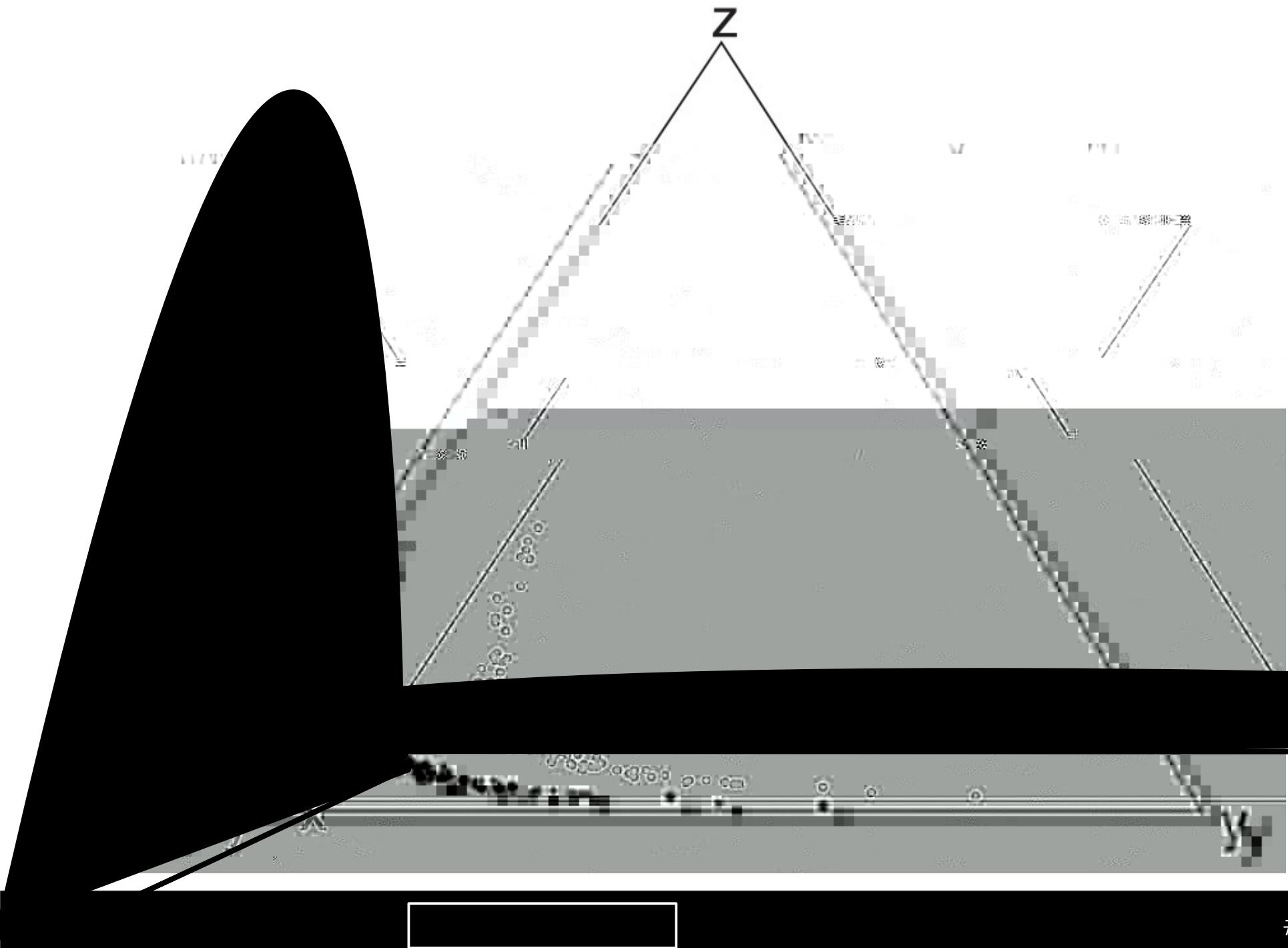


1. Isotopic measurements are compositional data
2. Error correlations are commonplace in geochronology









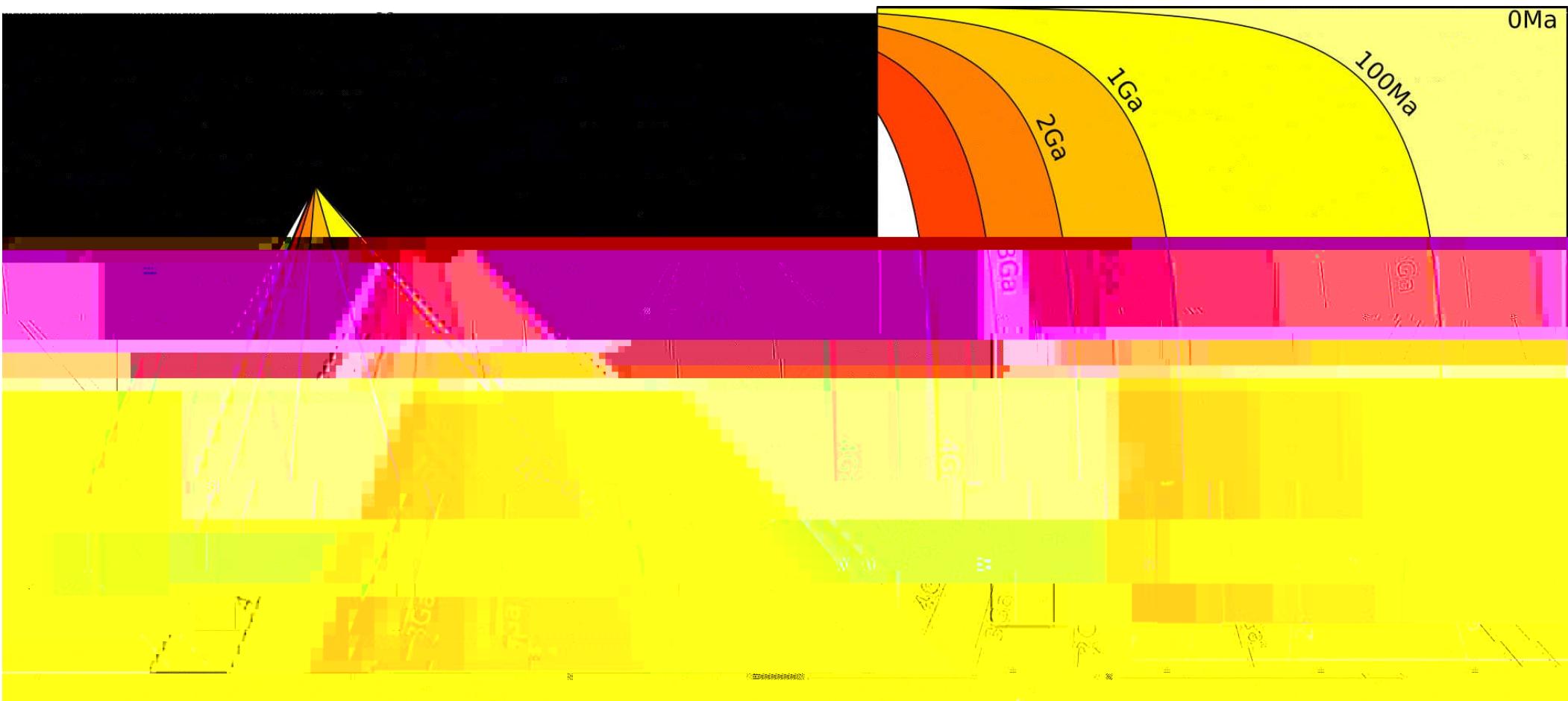
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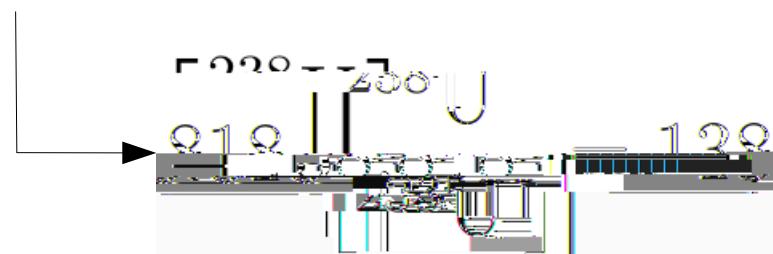
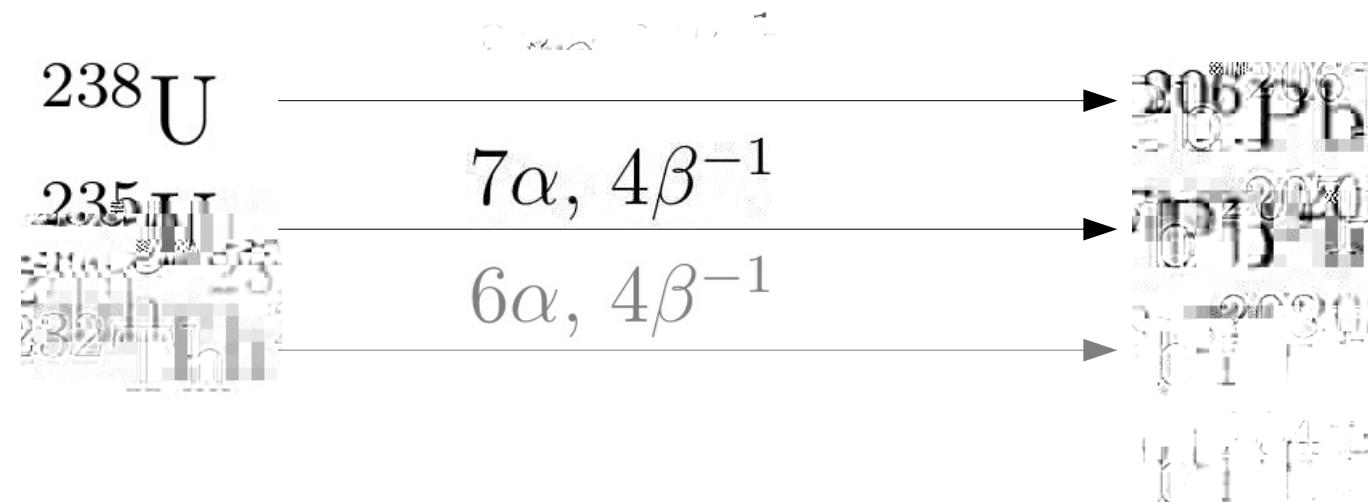




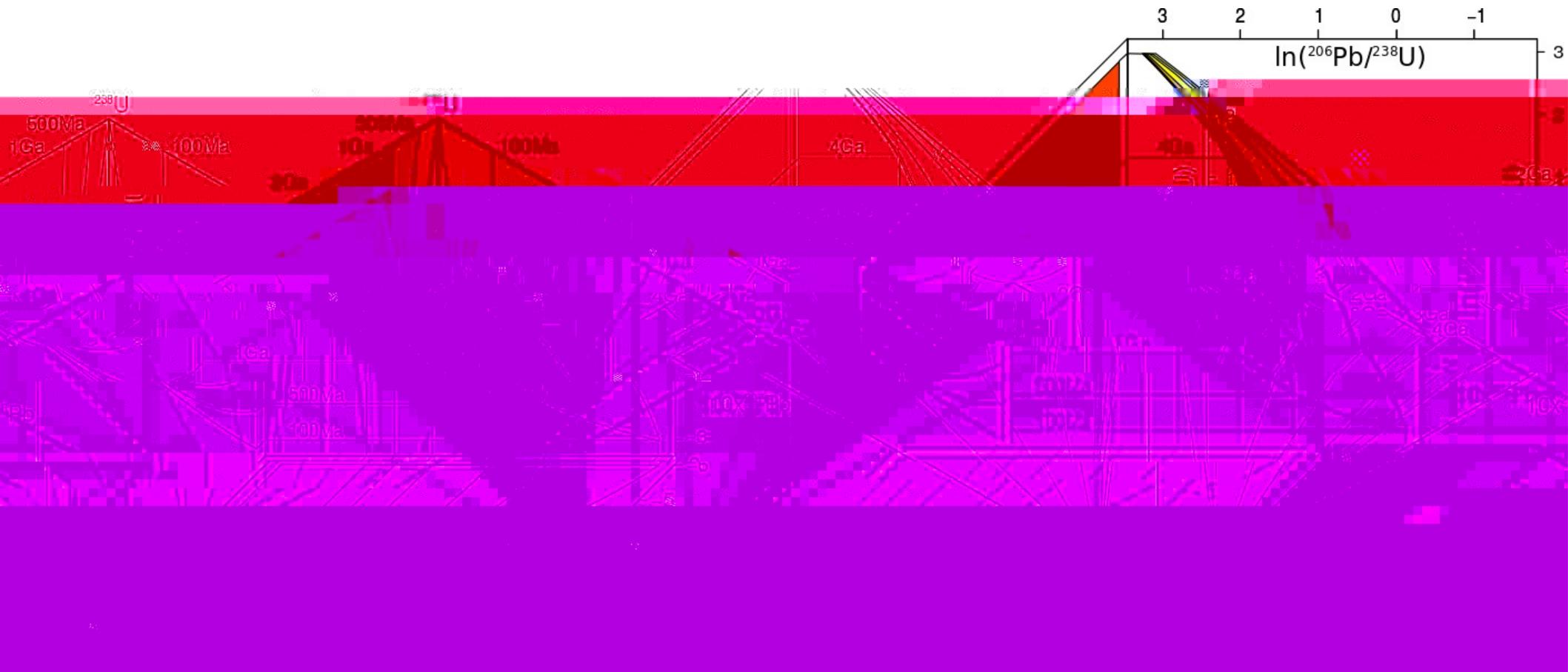


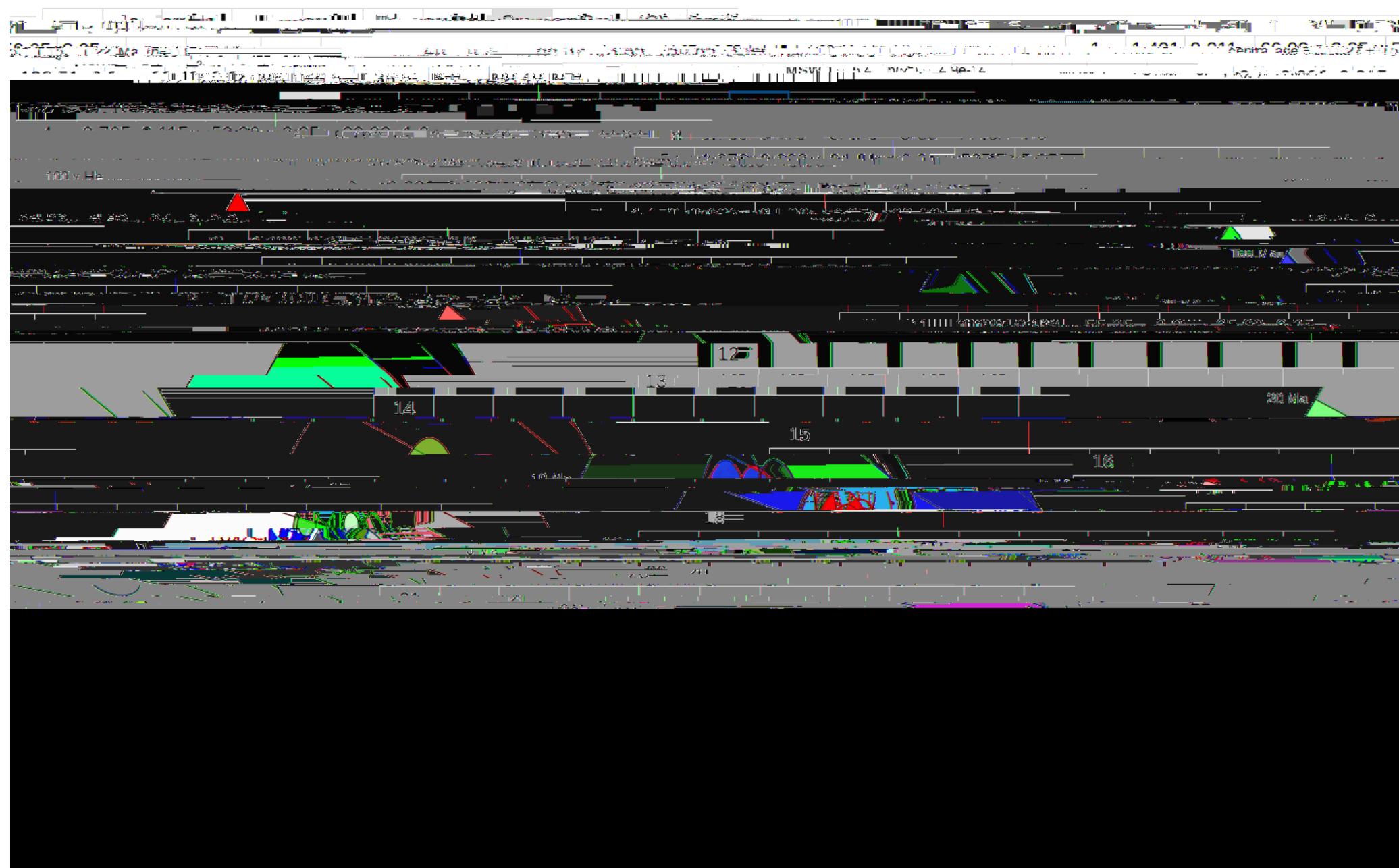


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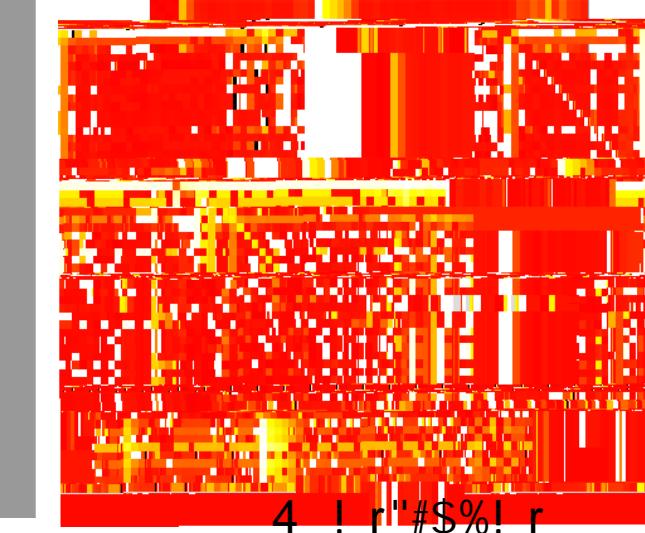
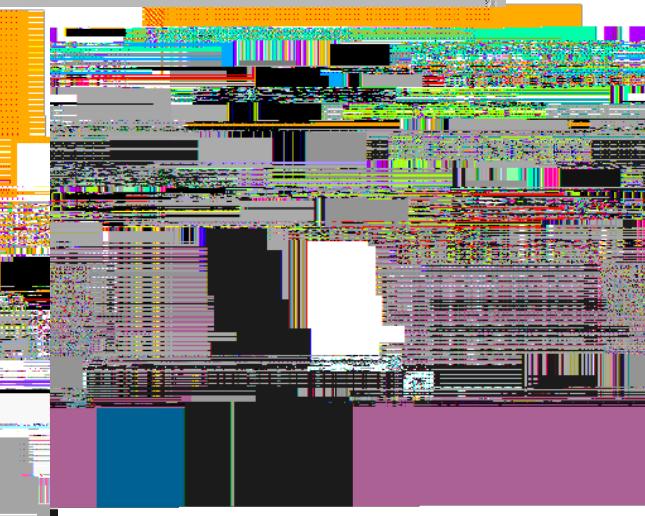
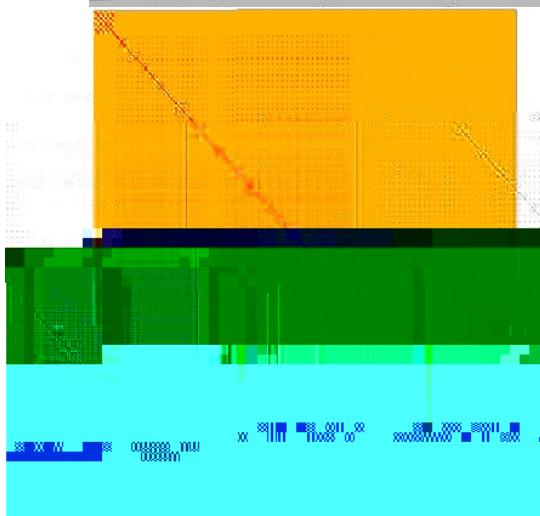
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1. Isotopic measurements are compositional data
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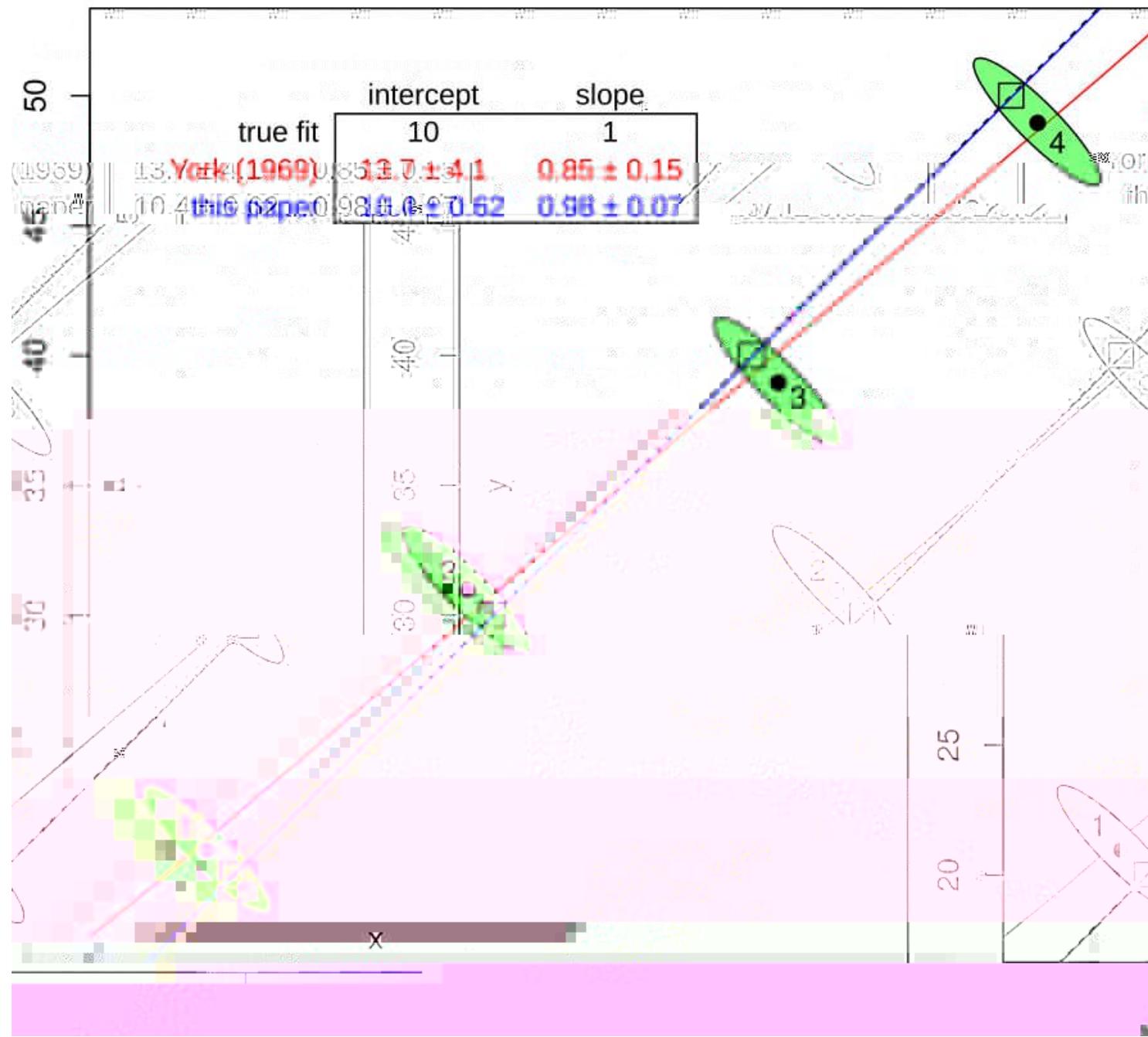


$$\sigma_z^{22} = \left[\begin{array}{cc} \frac{\partial f}{\partial x} & \frac{\partial f}{\partial y} \end{array} \right] \left[\begin{array}{c} \sigma_x^2 \\ \text{cov}(x,y) \end{array} \right] \left[\begin{array}{c} \text{cov}(x,y) \\ \sigma_y^2 \end{array} \right] \left[\begin{array}{cc} \frac{\partial f}{\partial x} & \frac{\partial f}{\partial y} \end{array} \right]$$



\$







\$'

